

INDIAN SCHOOL MUSCAT

CHEMISTRY DEPARTMENT

QUESTION BANK

Halo alkanes and Halo arenes

1 Define

- (i) Chiral carbon
- (ii) Chiral molecule
- (iii) Enantiomer
- (iv) Racemic mixture

2 Explain the following with suitable example

- (i) Racemisation
- (ii) Inversion of configuration
- (iii) Retention of configuration

3 Give reasons:

- (i) Thionyl chloride is best reagent for converting alcohols to haloalkanes.
- (ii) Alkyl halides have higher boiling points than corresponding hydrocarbons
- (iii) Benzylic and allylic halides follow S_N1 mechanism.
- (iv) H_3PO_4 is used with KI for iodination of alcohols.
- (v) Halogens are deactivating but 'o, p' directive.
- (vi) Presence of electron withdrawing groups on benzene ring increases tendency of S_N reaction.

- 4 Write equation to illustrate
- (i) Sandmeyer reaction
 - (ii) Finkelstein reaction
 - (iii) Swart's reaction
 - (iv) Wurtz reaction.
- 5 Explain with suitable example the following mechanism
- (i) S_N1
 - (ii) S_N2
 - (iii) S_N in aryl halides
- 6 What happens when:
- (i) Chlorobenzene is treated with Mg in ether and then reacted with water
 - (ii) Cyclohexene is treated with Br₂ /uv light
 - (iii) Ethylbromide is treated with excess ammonia
 - (iv) Isopropylbromide is treated with sodium ethoxide
 - (v) Methyl bromide is treated with metallic sodium in ether
- 7 Bring about the following conversions:
- (i) 2-methyl propene to 2-chloro-2-methylpropane
 - (ii) benzene to 4-bromo nitrobenzene
 - (iii) benzylalcohol to 2-phenylethanoic acid
 - (iv) ethanol to butane
 - (v) propene to 1-propanol
 - (vi) 1-bromopropane to 2-bromopropane
 - (vii) 2-bromopropane to 1-bromopropane

- (viii) Benzene to biphenyl
- (ix) Methylbromide to acetic acid
- (x) Tert- butylbromide to isobutylbromide

8 Give a test to distinguish

- (i) Ethyl chloride and phenyl chloride
- (ii) 1-Chloropropene and 3-chloropropene